

The Effect of 19-Substituent on the Stereospecific Hydroxylation of Δ^5 -Steroids Using Performic Acid as Oxidant

Wei Gang LU, Jing Yu SU, Long Mei ZENG*, Hong LI

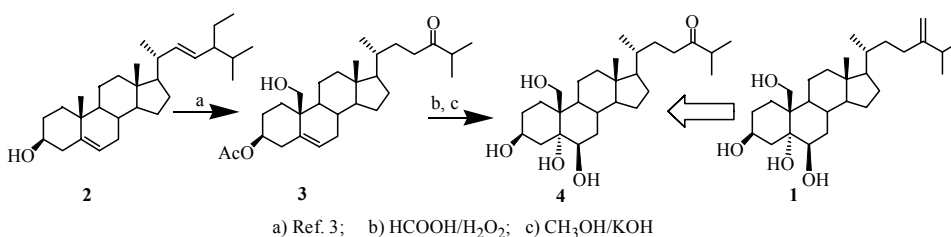
School of Chemistry and Chemical Engineering, Sun Yat-sen (zhongshan)
University, Guangzhou 510275

Abstract: In the studying of the stereospecific hydroxylation of different Δ^5 -steroids with performic acid, followed by hydrolysis with $\text{CH}_3\text{OH}/\text{KOH}$, we found 19-substituting groups considerably affected the reaction and we put forward the mechanism.

Keywords: Δ^5 -Steroids, performic acid, 19-substituting group, hydroxylation.

24-Methylenecholesta-3 β , 5 α , 6 β , 19-tetrol **1**, isolated from the soft corals, *Nephthea albida* and *N. tiexial verseveldt* by L. M. Zeng¹, showed significant cytotoxicity to A-549, H-29, KB and P-388 cell lines with ED_{50} values of 0.81, 0.93, 0.39 and 0.34 $\mu\text{g}/\text{mL}$, respectively². We have designed a synthetic route for the synthesis of **1** as shown in **Scheme 1**.

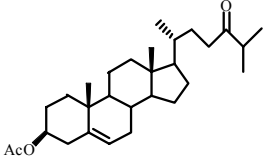
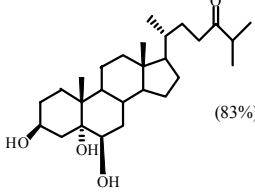
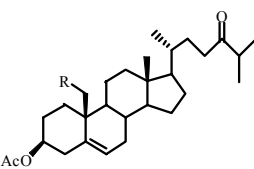
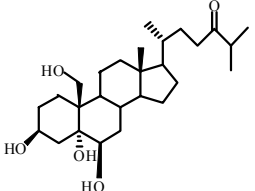
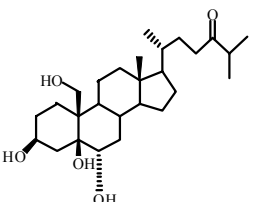
Scheme 1



In this route, the introduction of keto-containing side-chain and 19-hydroxy group was completed in 8 steps starting from stigmaterol **2** referring to the literature³. In the hydroxylation of 5, 6-olefin **3** to 5 α , 6 β -diol **4** we used performic acid as epoxidizing agent, followed by hydrolysis with $\text{CH}_3\text{OH}/\text{KOH}$. During the study of this conversion we found the steroid without 19-substituting group provided the expected product exclusively, while the steroid with 19-hydroxy or 19-acetoxy group provided two products in a relatively low yield.

* E-mail: ceszlm@zsu.edu.cn

Table 1 Hydroxylation of Δ^5 -steroids with performic acid as oxidant

Substrate	Products (yield)	
 5	 6 (83%)	
 3 R = HO 3a R = CH ₃ COO	 4 (41%) 4 (21%)	 7 (15%) 7 (20%)

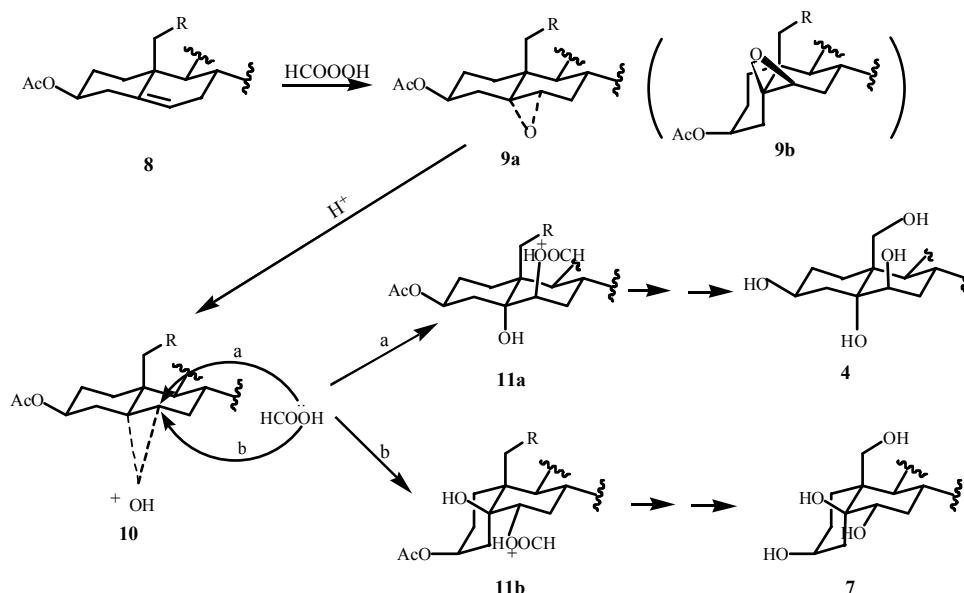
As shown in **Table 1**, the Δ^5 -steroidal hydroxylation of compound **5** by employing J. Romo's method⁴ with performic acid, followed by hydrolysis with CH₃OH/KOH, gave compound **6** in a yield comparable to the literature. However, 19-substituted compounds **3** and **3a**, subjected to the similar situation, gave two products **4** and **7** (tracked by TLC and separated by silica gel chromatography, benzene/acetone=1/1) in a relatively low yield.

It is interesting that these two compounds (**4** and **7**) have the same molecular formula C₂₇H₄₆O₅ (determined by Micro-analysis and FAB-mass spectrometry) and all have one primary, two secondary and one tertiary alcohols (determined by ¹H NMR, ¹³C NMR and DEPT). According to their NMR spectral data^{5,6} and literature survey^{1,7}, we confirmed these two compounds being configurational isomers, cholest-24-oxo-3 β , 5 α , 6 β , 19-tetrol **4** and cholest-24-oxo-3 β , 5 β , 6 α , 19-tetrol **7**.

We suggested that the steric hindrance of 19-substituting groups might be the explanation, here we proposed the mechanism for the formation of compound **4** and **7** (**Scheme 2**).

The reaction of performic acid with Δ^5 -steroids **8** produces the 5 α , 6 α - and 5 β , 6 β -epoxides (**9a** and **9b**) with the epoxidation taking place predominantly from the α -face of the molecule⁸. The protonation of the α -epoxide **9a** will lead to the formation of oxonium intermediate **10**, which are susceptible to nucleophilic attack. Intermediate **10** is expected to be particularly reactive at its relatively unhindered 6-position by nucleophile HCOOH⁹ either from the α -face or the β -face, to lead to two products, **11a** and **11b**. The hydrolysis of **11a** and **11b** with CH₃OH/KOH produces **4** and **7** respectively. If R is hydrogen, the approach of the nucleophile in the intermediate **10** is preferred from the β -face because of the *trans*-A/B ring system unchanged. If R is hydroxy or acetoxy group, steric factor suggested that the attack from the β -face might be sterically restricted, while the attack from the α -face will lead to the formation of compound **7**, which has a relatively unstable *cis*-A/B ring system.

Scheme 2



Steric effect of 19-substituting group on the cleavage of epoxide, only rings A and B are shown

In conclusion, the existence of R group increases the steric hindrance on the β -face, and it reduces the approaching of nucleophile from the β -face to the extent that the approaching from the α -face may not be neglected.

Acknowledgment

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References and Notes

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5. The physical data of compound 4: Anal. Calcd. for $\text{C}_{27}\text{H}_{46}\text{O}_5$: (%) C 72.06, H 10.08 Found: C 71.54, H 10.23; FAB-MS m/z 451 ($\text{M}+\text{H}$)⁺; IR (KBr, cm^{-1}) 3327, 2941, 2869, 1707, 1464, 1380, 1047, 976; ¹H-NMR (DMSO- d_6 , ppm) 0.66 (s, 3H, H-18), 0.86 (d, 3H, $J = 6.5\text{Hz}$, H-21), 0.99 (d, 6H, $J = 7.0\text{Hz}$, H-26 and H-27), 2.60 (m, 1H, $J = 7.0\text{Hz}$, H-25), 3.27 (m, 1H, H-6), 3.52 (dd, 1H, $J = 12.0, 6.0\text{Hz}$, H_a-19), 3.69 (s, 1H, 5 α -OH), 3.85 (m, 1H, H-3), 3.99 (dd, 1H, $J = 12.0, 3.5\text{Hz}$, H_b-19), 4.11 (d, 1H, $J = 5.5\text{Hz}$, 3-OH), 4.50 (m, 1H, 19-OH), 5.02 (dd, 1H, $J = 5.0, 1.5\text{Hz}$, 6-OH); ¹³C-NMR (Pyridine- d_5 , ppm) 12.8 (CH₃), 18.4 (CH₃), 18.5 (CH₃), 18.7 (CH₃), 22.5 (CH₂), 24.5 (CH₂), 28.5 (CH₂), 29.7 (CH₂), 30.3 (CH₂), 32.4 (CH), 32.6 (CH₂), 35.0 (CH₂), 35.8 (CH), 37.4 (CH₂), 40.9 (CH), 41.2 (CH₂), 42.9 (C), 43.5 (CH₂), 43.8 (C), 46.4 (CH), 56.4 (CH), 57.6 (CH), 64.7 (CH₂), 67.4 (CH), 75.8 (CH), 76.2 (C), 214.2 (C).

6. The physical data of compound 7: Anal. Calcd. for C₂₇H₄₆O₅: (%) C 72.06, H 10.08 Found: C 71.34, H 10.40; FAB-MS *m/z* 451 (M+H)⁺; IR (KBr, cm⁻¹) 3409, 2937, 2871, 1709, 1463, 1381, 1047, 988; ¹H-NMR (DMSO-d₆, ppm) 0.61 (s, 3H, H-18), 0.85 (d, 3H, *J* = 6.5Hz, H-21), 0.99 (d, 6H, *J* = 7.0Hz, H-26 and H-27), 2.60 (m, 1H, *J* = 7.0Hz, H-25), 3.48 (dd, 1H, *J* = 11.0, 6.0Hz, H_a-19), 3.73 (dt, 1H, *J* = 11.5, 4.5Hz, H-6), 3.83 (dd, 1H, *J* = 11.0, 3.5Hz, H_b-19); 3.98 (brs, half width 8.5Hz, 1H, H-3); 4.04 (d, 1H, *J* = 4.0Hz, 3-OH), 4.67 (dd, 1H, *J* = 6.0, 3.5Hz, 19-OH), 5.11 (d, 1H, *J* = 5.0Hz, 6-OH), 5.33 (s, 1H, 5-OH); ¹³C-NMR (Pyridine-d₅, ppm) 12.2 (CH₃), 18.4 (CH₃), 18.5 (CH₃), 18.7 (CH₃), 20.6 (CH₂), 21.9 (CH₂), 24.4 (CH₂), 28.3 (CH₂), 28.4 (CH₂), 30.2 (CH₂), 30.9 (CH₂), 34.2 (CH), 35.6 (CH), 36.8 (CH₂), 37.4 (CH₂), 40.5 (CH₂), 40.9 (CH), 42.9 (C), 43.2 (CH), 43.9 (C), 56.1 (CH), 57.1 (CH), 65.7 (CH₂), 67.0 (CH), 72.1 (CH), 81.6 (C), 214.2 (C).
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